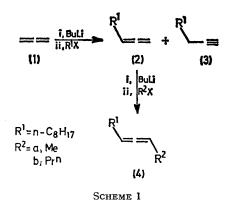
Simple Route to Mono-, Di-, and Tri-substituted Allenic Compounds

By Gérard Linstrumelle and Didier Michelot

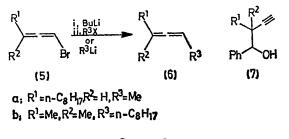
(E.R. 12 C.N.R.S.; Laboratoire de Chimie, Ecole Normale Supérieure, 24, rue Lhomond, 75231 Paris Cedex 05)

Summary Allenic compounds can be prepared in high yields without appreciable rearrangement by alkylation of allenic lithium reagents.

DOUBLE bonds have been synthesised stereospecifically by alkylation of vinylic lithium reagents.¹ We report a high yield synthesis of mono-, di-, and tri-substituted allenes *via* allenic lithium reagents derived from terminal allenichydrocarbons or from bromoallenes. Although classically



one might expect the formation of both allenic and acetylenic products corresponding to the allenic anion and the related acetylenic species, the formation of acetylenic compounds was of little importance in the present case.



Scheme 2

Propa-1,2-diene (1) was lithiated in tetrahydrofuran at -70° with BuⁿLi² (1 equiv.). Reaction with octyl iodide or bromide (1 equiv.) gave 86% of a mixture (87:13) of the allene (2) and the acetylene (3).³ The allene (2) was easily purified by treatment with aqueous AgNO₃ and was identical to an authentic sample;^{4,5} M^+ 152; δ (CCl₄) 4·47–5·30 (3H, m).

Lithiation of the monosubstituted allene (2) occurs at the terminal position. A convenient procedure for the synthesis of 1,3-disubstituted allenes has been developed: for example, undeca-1,2-diene (2) was treated sequentially with equimolecular quantities of BunLi and MeI to provide dodeca-2,3-diene (4a) in 93% yield (94% purity); M^+ 166; ν 1964 cm^{-1}; δ (CCl₄) 1.55–1.73 (3H, m) and 4.72–5.26 (2H, m). BunLi and PrnI similarly gave (96%) tetradeca-4,5-diene (4b) (95% purity), b.p. 130—140 °C at 15 mmHg; M^+ 194; ν 1963 cm⁻¹; δ (CCl₄) 4·81—5·18 (2H, m).

Allenic lithium reagents could also be obtained from bromoallenes by the action of Li or BunLi.⁶ Treatment of 1-bromoundeca-1,2-diene (5a) (prepared from undec-1-yn-3-ol according to a known procedure⁷) with BuⁿLi (1 equiv.) in ether at -70° followed by addition of tetrahydrofuran and MeI (1 equiv.) gave (90%) dodeca-2,3-diene (6a) (purity 92%). The bromoallene (5b),⁷ under the same conditions, provided (91%) the pure trisubstituted allene (6b), b.p. 65 °C at 0.1 mmHg; M^+ 180; v 1970 cm⁻¹; δ (CCl₄) 1.64 (6H, d, J 3 Hz) and 4.87 (1H, m).

The allene (6b) could also be generated (65%) directly by action of an alkyl-lithium upon the bromoallene (5b) (via halogen-metal interconversion and coupling).

Except in the case $(1) \rightarrow (2) + (3)$ where 13% of the acetylene (3) was formed, the allenic compounds described here were generated without significant isomerization (<5%) to the acetylenic isomers.

The reactivity of these allenic lithium reagents with alkyl halides contrasts with their reactivity with carbonyl compounds; the acetylenic alcohols (7a) and (7b) were formed in good yield (70 and 59%) without noticeable amounts of the allenic alcohols.2,8

The authors gratefully acknowledge financial support from the Délégation Générale à la Recherche Scientifique et Technique and the Centre National de la Recherche Scientifique, and thank Dr. Sylvestre Julia for interest and encouragement.

(Received, 15th April 1975; Com. 436.)

¹G. Linstrumelle, Tetrahedron Letters, 1974, 3809; J. Millon, R. Lorne, and G. Linstrumelle, Synthesis, 1975, in the press. ² F. Jaffé, J. Organometallic Chem., 1970, 23, 53.

³ Separated by g.l.c.; physical data are in good agreement with the structure assigned; M. Larchevêque, Bull. Soc. chim. France, 1968, 3387.

⁴ W. R. Moore and H. R. Ward, J. Org. Chem., 1962, 27, 4179.

⁵ Prepared by the method of M. Gaudemar, Ann. Chim. (Paris), 1956, 1, 161.

⁶ Propa-1,2-dienyl-lithium has been obtained from 1-bromopropa-1,2-diene by the action of lithium or methyl-lithium, M. Le Quan, personal communication.

⁷ S. R. Landor, A. N. Patel, P. F. Whiter, and P. M. Greaves, J. Chem. Soc. (C), 1966, 1223.
⁸ However, formation of allenic alcohols has been reported: T. P. Patrick, E. C. Haynie, and W. J. Probst, J. Org. Chem. 1972, 37, 1553; H. G. Viehe, 'Chemistry of Acetylenes,' Marcel Dekker, New York, 1969, p. 407.